

### SCIENCE LESSON WORKSHEET

### **DECODING MOLECULAR FORMULAS**

Molecular formulas tell us what elements and atoms are in a molecule. Learn how to decode them with help from the Periodic Table of Elements.

### HOW TO DECODE A MOLECULAR FORMULA

Use these steps to figure out what each molecules below are made of.

Take a look at the CO2 sample below, then move on to the next page.

Step 1: Underline the Capital Letters. This will tell you how many elements are in your molecule.

Step 2: Symbols-Write the symbols the molecule has in the symbol column.

Step 3: Element Name-Use the periodic table to identify what elements the symbol represents and write the down in the element column.

Step 4: How many atoms? - Write the number of atoms of each element in the How many atoms column. If there is a small number beside the symbol, this indicates how many atoms of this element there are in the molecule. If there is NO small

# Molecular Formula: CO2 (Carbon Dioxide)

| Symbol | Element Name | How many atoms? |
|--------|--------------|-----------------|
| С      | Carbon       | 1               |
|        |              |                 |
|        |              |                 |
| 0      | Oxygen       | 2               |
|        |              |                 |
|        |              |                 |



Use the Periodic Table of Elements to decode the following molecules.

## Molecular Formula: NH3 (Ammonia)

| Symbol | Element Name | How many atoms? |
|--------|--------------|-----------------|
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |

# Molecular Formula: CH4 (Methane)

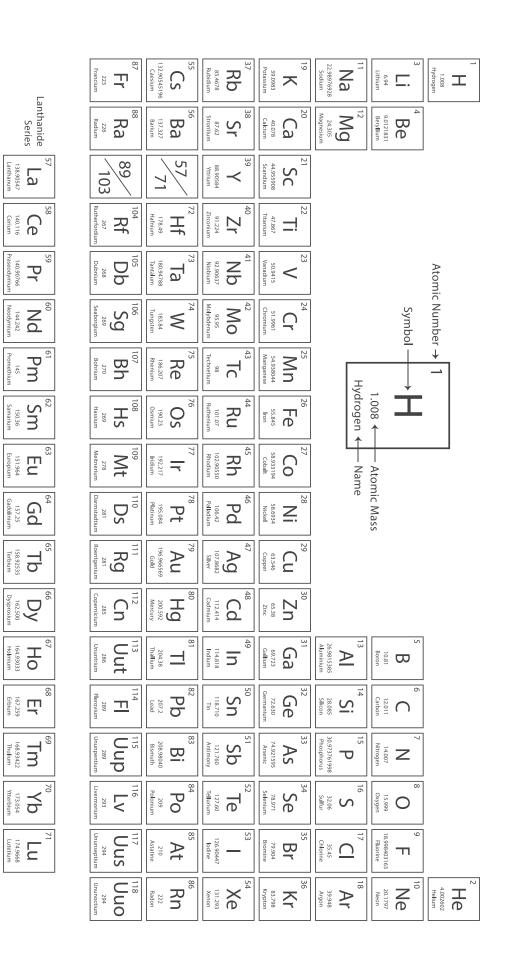
| Symbol | Element Name | How many atoms? |
|--------|--------------|-----------------|
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |

# Molecular Formula: H2SO4 (Sulfuric Acid)

| Symbol | Element Name | How many atoms? |
|--------|--------------|-----------------|
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |
|        |              |                 |



# NAME THAT MOLECULE CHALLENGE





Lanthanide

Series

La

140.116 Cerium

PΥ

138.90547 Lanthanum

140.90766 raseodymiu

144.242 Neodymiu Nd

Actinide Series

Ac

Pa

Z p

Pu

Am

Cm

燢

**ES** 

F<sub>0</sub>

Md

**N**0

103 LT

266

\*\*wrencium\*\*

162.500 Dysprosiur Dy

164.93033 Holmium

Но

167.259 Erbium

Lutetium